

Methyl *N*-(benzyloxycarbonyl)oxamateRaimund Marx,<sup>a</sup> Steffen J.  
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bolte@chemie.uni-frankfurt.deThe title compound, C<sub>11</sub>H<sub>11</sub>NO<sub>5</sub>, is almost planar. The  
molecules pack as hydrogen-bonded dimers.Received 20 August 2003  
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## Comment

N- and C-protected ethyl oxamates are excellent nitrogen  
nucleophiles in Gabriel reactions (Berrée *et al.*, 1999). They  
are available by oxidation of protected serine or threonine  
(Stachulski, 1982). For our synthesis of methyl *N*-(benzyl-  
oxycarbonyl)oxamate, (I), we chose oxidation with RuO<sub>2</sub>/  
NaIO<sub>4</sub> (Garner & Park, 1990), which leads to better yields.

## Key indicators

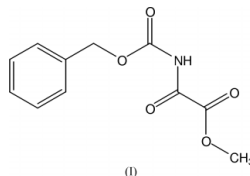
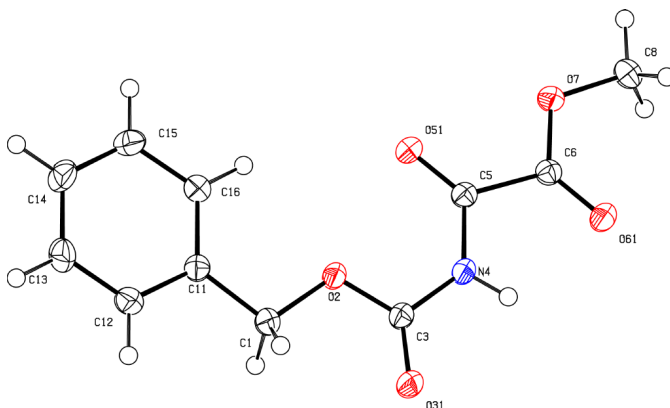
Single-crystal X-ray study  
*T* = 173 K  
Mean  $\sigma$ (C–C) = 0.002 Å  
*R* factor = 0.042  
*wR* factor = 0.098  
Data-to-parameter ratio = 19.3For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.The title compound is almost planar (r.m.s. deviation for all  
non-H atoms is 0.06 Å). All torsion angles excluding H atoms  
are close to either 0 or 180°. The biggest deviations from  
planarity are –2.75 (18)° for O2–C3–N4–C5 and  
175.02 (10)° for C11–C1–O2–C3. An interesting feature of  
(I) is the rather long *Csp*<sup>3</sup>–*Csp*<sup>3</sup> bond between atoms C5 and  
C6. The molecules pack as hydrogen-bonded dimers.Only one similar compound, (II) (Wyss & Brisse, 1984), has  
been found in the Cambridge Structural Database (CSD,  
Version 5.24 of April 2003; Allen, 2002) which differs from (I)  
in the terminal groups: the benzyl group in (I) is replaced by a  
*tert*-butyl group in (II) and the methyl group in (I) is replaced  
by a benzyl group in (II). Although this leads to a completely

Figure 1

Perspective view of the title compound with the atom numbering;  
displacement ellipsoids are drawn at the 50% probability level.

different conformation of (II), the latter also packs as hydrogen-bonded dimers. The rather long  $Csp^3-Csp^3$  (1.533 Å) bond is also observed in (II).

### Experimental

*N*-Benzyloxycarbonyl-L-serine methyl ester (2 g) was dissolved in 200 ml of a 3:2 mixture of acetone and water. With stirring, 10 equivalents of sodium periodate and 0.08 equivalents of ruthenium dioxide were added. The suspension was stirred for 24 h at room temperature. 10 ml of 2-propanol were added and the mixture stirred for a further 90 min to consume excess oxidant. The suspension was separated by centrifugation, yielding a clear yellow solution which was then concentrated *in vacuo*. HPLC purification yielded 1.44 g of pure *N*-benzyloxycarbonyl methyl oxamate (77%). The crystals were obtained directly from the HPLC fractions.

#### Crystal data

$C_{11}H_{11}NO_5$	$D_x = 1.458 \text{ Mg m}^{-3}$
$M_r = 237.21$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 512 reflections
$a = 8.0139 (8) \text{ \AA}$	$\theta = 3.2-20.7^\circ$
$b = 5.4828 (5) \text{ \AA}$	$\mu = 0.12 \text{ mm}^{-1}$
$c = 24.701 (3) \text{ \AA}$	$T = 173 (2) \text{ K}$
$\beta = 95.446 (2)^\circ$	Plate, colourless
$V = 1080.4 (2) \text{ \AA}^3$	$0.52 \times 0.28 \times 0.12 \text{ mm}$
$Z = 4$	

#### Data collection

Siemens 1K CCD three-circle diffractometer	2248 reflections with $I > 2\sigma(I)$
$\omega$ scans	$R_{\text{int}} = 0.042$
Absorption correction: none	$\theta_{\text{max}} = 30.6^\circ$
17271 measured reflections	$h = -11 \rightarrow 11$
3064 independent reflections	$k = -7 \rightarrow 7$
	$l = -33 \rightarrow 34$

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0434P)^2 + 0.2657P]$
$R[F^2 > 2\sigma(F^2)] = 0.042$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.098$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.02$	$\Delta\rho_{\text{max}} = 0.36 \text{ e \AA}^{-3}$
3064 reflections	$\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$
159 parameters	Extinction correction: <i>SHELXL97</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.0114 (17)

**Table 1**

Selected geometric parameters (Å).

C1—O2	1.4529 (15)	C5—O51	1.2057 (14)
O2—C3	1.3243 (14)	C5—C6	1.5430 (17)
C3—O31	1.2088 (15)	C6—O61	1.1980 (14)
C3—N4	1.3868 (16)	C6—O7	1.3221 (15)
N4—C5	1.3737 (16)	O7—C8	1.4570 (15)

**Table 2**

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N4-H4 \cdots O31^i$	0.863 (17)	2.156 (17)	2.9778 (14)	158.9 (14)

Symmetry code: (i)  $-x, -y, 1-z$ .

H atoms bonded to C atoms were refined with fixed individual displacement parameters [ $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$  or  $1.5 U_{\text{eq}}(\text{C}_{\text{methyl}})$ ] using a riding model with  $C-H = 0.99 \text{ \AA}$  or methyl  $C-H = 0.98 \text{ \AA}$ . The H atom bonded to N was refined freely.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 1990); software used to prepare material for publication: *SHELXL97*.

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